RESEARCH ARTICLE

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Screening freshness of seafood by measuring trimethylamine (TMA) levels using helium-plasma ionization mass spectrometry (HePI-MS)



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Abstract

Background: Trimethylamine (TMA) is a marker used for monitoring the quality of seafood because it is the primary component of the "fishy" odor.

Methods: The levels of TMA in seafood samples were directly measured by helium-plasma ionization mass spectrometry (HePl-MS). Each sample was directly exposed to the HePl source, and the intensity of the m/z 60 signal for protonated TMA was monitored by a selected-ion-recording (SIR) protocol. Using a set of TMA-spiked water standards, the TMA levels in seafood samples were quantified.

Results: The signal intensity of the m/z 60 ion from shrimp samples maintained at room temperature for 2 days can be attenuated to baseline levels by adding lime juice. The amounts of TMA in samples of salmon and shrimp recovered from some sushi preparations, and in squid samples, were found to be 0.24 μ g, 0.16 μ g, and 17.2 μ g per gram, respectively.

Conclusions: HePI-MS is an efficient technique to screen and monitor the TMA content and assess the quality of seafood.

Keywords: Trimethylamine (TMA), Helium-plasma ionization, HePI, Ambient mass spectrometry, Seafood quality, Fish odor, Screening freshness of seafood

Introduction

Fresh seafood is highly perishable and therefore requires utmost care during processing, transportation, and storage in order to prevent decomposition. Approximately 200 million metric tons of seafood is directly consumed by people globally per annum as reported by the UN Food and Agricultural Organization. Despite the high perishability, many connoisseurs prefer consuming seafood preparations raw or only lightly preserved. As a result, the market has recently seen an increasing demand for fresh seafood. Thus, in order to meet consumer demands and comply with legislative regulations, a quality assessment performed on the product before it is offered

to the consumers is of paramount importance. In seafood, biogenic amines are formed upon storage thorough enzymatic and microbial action on amino-acids. Among the biogenic amines, trimethylamine (TMA) is the primary component that imparts the fishy odor (Bedia Erim 2013). Thus, TMA is commonly used as a marker to qualitatively and semi-quantitatively detect the spoilage of fish (Oetjen and Karl 1999; Pedrosa-Menabrito and Regenstein 1990; Timm and Jørgensen 2002). TMA is produced by the oxidation of choline by bacteria in marine animals by TMA-lyase. TMA also accumulates by the reduction of trimethylamine N-oxide (TMAO) by the enzyme TMAO reductase in the tissues of decaying marine animals. TMA is toxic to humans: it is oxidized in the liver to form TMAO (Seibel and Walsh 2002), which has been recognized as an agent that causes cardiovascular disease (Landfald et al. 2017).

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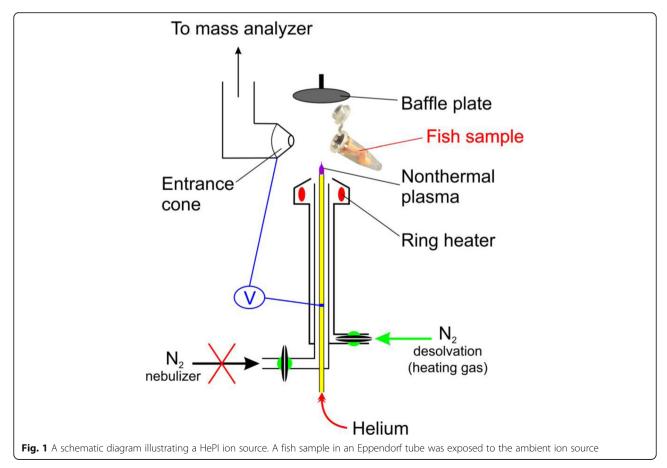
Several sensory- and instrument-based techniques are available to monitor the quality of seafood. Most of these methods rely on the detection of TMA, which is one of the main compounds responsible for the malodor of poor-quality seafood (Oetjen and Karl 1999; Timm and Jørgensen 2002). The correlation between extracted TMA and the age and quality of seafood has been well-demonstrated (Malle et al. 1996; Malle and Tao 1987; Oetjen and Karl 1999; Romero-González et al. 2012; Timm and Jørgensen 2002).

More elaborate instrumental methods have evolved through the years. At present, gas chromatography (Namieśnik et al. 2003; Shim and Baek 2012), solid-phase micro-extraction (Chan et al. 2016; Shim and Baek 2012), or solvent extraction (daCosta et al. 1990; Oetjen and Karl 1999), ion mobility (Bota and Harrington 2006; Cheng et al. 2017), nuclear magnetic resonance spectroscopy (Podadera et al. 2005), ion chromatography (Erupe et al. 2010; Li et al. 2009), capillary electrophoresis (Li and Lee 2007; Timm and Jørgensen 2002), high-resolution rotational terahertz (THz) spectroscopy (Hindle et al. 2018), and high-performance liquid chromatography methods (Cháfer-Pericás et al. 2004; Hyötyläinen et al. 2001; Romero-González et al. 2012) are the most widely adopted techniques to measure the amounts of primary and

secondary amines in various matrices. The major advantages of these chromatographic techniques are higher sensitivity, specificity, and ability to determine several substances simultaneously.

A drawback of many traditional analytical techniques employed to determine the quality of seafood is the time-consuming and laborious TMA extraction step, and the difficulty in handling low-molecular-mass amines due to their high water solubility and volatility. The technique we have developed—based on ambient-pressure helium-plasma ionization mass spectrometry (HePI-MS)—does not require the TMA extraction step or chromatographic separation.

In this study, we employed ambient-pressure HePI-MS (Yang and Attygalle 2011) to screen the freshness of seafood, because the technique affords a *direct* measurement of TMA levels in samples, without the need to resort to chemical extraction, or any other prior sample preparation. HePI is a versatile ambient-ionization MS technique, applicable to the analysis of a wide variety of samples, both organic and inorganic. It has been applied, for example, to the analysis of energetic materials (Yang et al. 2012), pharmaceutical preparations (Attygalle et al. 2014a, 2014b, Xia et al. 2016), halobenzenes (Attygalle et al. 2014a, 2014b; Gangam et al. 2015), phenolics and



quinones (Hassan et al. 2017), inorganic nitrates (Pavlov and Attygalle 2013), and inorganic mercury compounds (Weerasinghe et al. 2014). A major advantage of HePI is that it is highly portable and adaptable: any mass spectrometer with an electrospray ion source can be transformed into an ambient HePI instrument with ease, and no extensive hardware modifications are necessary (see Experimental Section). In addition, unlike other helium-

mediated sources such as Direct Analysis in Real Time (DART) and Flowing Atmospheric-Pressure Afterglow (FAPA), HePI is extremely economical in its helium consumption. Another important feature of HePI is that if a sample is sufficiently volatile or can be volatilized, it can be detected without any significant interference from the sample matrix. In this study, we investigated the capabilities of HePI to measure TMA levels in several seafood

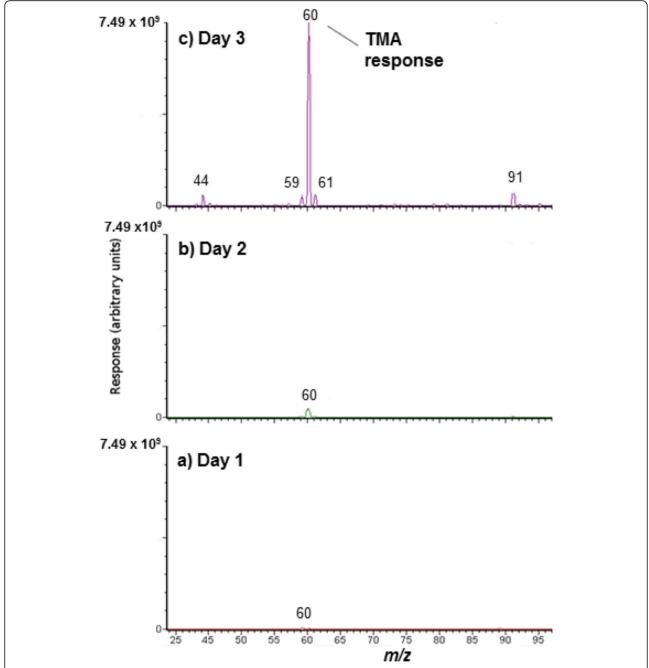


Fig. 2 Helium-plasma ionization mass spectra (*m/z* 20–100) recorded from headspace volatiles emanating from a shrimp sample kept at room temperature for one (a), two (b), and three days (c)

samples from various specimens of fish at different time points of storage at room temperature (0 to 96 h). Additionally, the reduction of the amount of free TMA by the addition of lime juice, similar to the action of many other common acidic seafood condiments (e.g., lemon juice and tartar sauce), was demonstrated.

Methods

Materials and sample preparation

High-purity helium (99.999%, Airgas, Radnor, PA) was used for all experiments. Trimethylamine was purchased from Sigma-Aldrich Co. (St. Louis, MO, USA). Concentrated hydrochloric acid and NaHCO₃ were purchased from Fisher Scientific (Hampton, NH). Lime juice

(ReaLime® 100%) and samples of seafood [cod, char, salmon, and squid (mantle only, not the tentacles)], displayed on ice, were purchased from a local store (Wegmans, Woodbridge, NJ) and transported to the laboratory on a bed of ice. Similarly, samples of shrimpand salmon-sushi, and fresh squid samples were purchased from the same store, and a sample of fresh shrimp was obtained from another local store (99 Ranch Market, Jersey City, NJ). From each seafood, eighteen representative samples (25 mm × 3 mm x 3 mm; 0.8 g each) were separated and placed in Eppendorf tubes (2.0 mL), which were kept at room temperature with lids tightly closed. For shrimp, the representative samples were cut from the region immediately below the head,

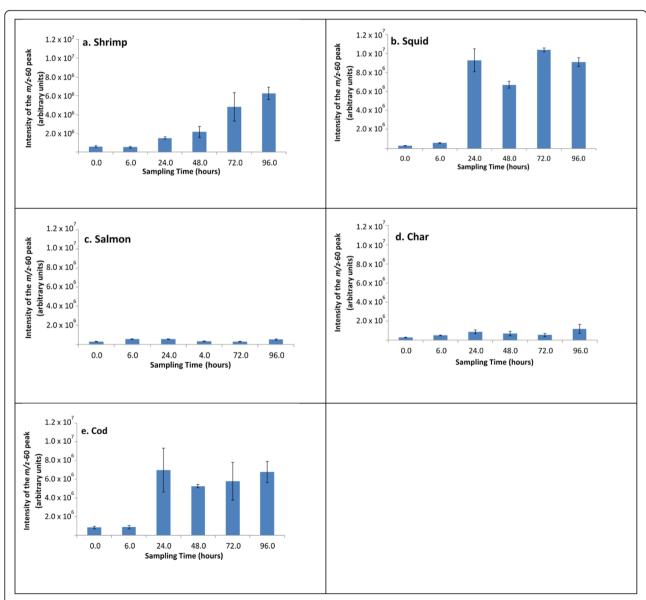
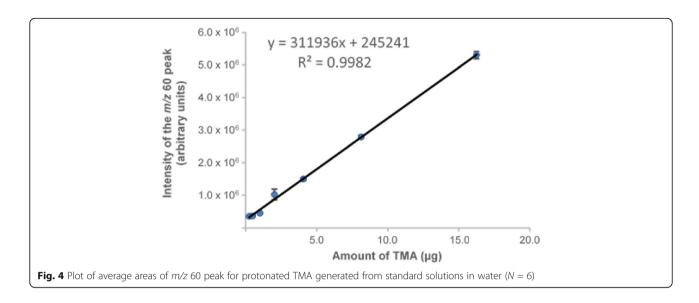


Fig. 3 Intensity of the m/z 60 signal for protonated TMA recorded from headspace volatiles of shrimp (**a**), squid (**b**), salmon (**c**), char (**d**), and cod (**e**) samples (N = 6) kept at room temperature for 96 h.



and for the sushi samples, samples were prepared only from the meat portion.

Mass spectrometric analysis

A stream (25–30 mL/min) of high-purity helium (99.999%, Airgas, Radnor, PA, USA) was passed through a metal capillary (100 μm ID) held at a high voltage (typically, 3 kV) for it to function as a helium-plasma ionization source, as described previously (Yang and Attygalle 2011). The capillary was placed at a 90° angle, about 10 mm away from the outer cone orifice of a single-quadrupole mass spectrometer (ZQ model, Waters Corp., Milford, MA, USA). Positive-ion mass spectra were recorded from headspace volatiles emanating from each sample (Fig. 1).

Prior to analysis, each Eppendorf tube containing a sample was opened and attached to the inner side of the ion-source glass cover. Mass spectra were acquired (m/z 25–200) at a rate of 2 scans per second, and the results were processed using MassLynx 4.0 software (Waters Corp., Milford, MA, USA). The cone voltage was held at 15 V; the source and desolvation gas temperatures were both maintained at 110 °C. After each analysis, the ion source was cleaned thoroughly by wiping it with a cotton swab soaked in methanol, and a background check was made to ensure that the m/z 60 signal intensity had returned back to the background level before a new sample was introduced.

Three samples of each seafood (cod, char, salmon, shrimp, and squid), maintained inside Eppendorf tubes at room temperature, were analyzed at 0, 6, 24, 48, 72, and 96 h. Each tube containing a sample was opened and placed immediately in the HePI source. The tube neck was positioned at a pre-determined fixed point at 1.0 cm distance from both the entrance-cone orifice and

the HePI plasma flame. The amount of TMA emanating from each sample was recorded for 0.5 min by a selected-ion-recording (SIR) experiment monitoring the abundances of the m/z 60 and 61 ions (dwell time 0.1 s; inter-scan delay 0.1 s). Each sample was analyzed in duplicate by the SIR procedure. Then, the average m/z 60 peak intensity and that of m/z 61 were calculated for each 0.5 min acquisition period.

Trimethylamine standard curve

The concentration of the commercial TMA solution was determined to be 3.5 M, by titrating it against a 0.12 M standard HCl solution, using methyl orange as the indicator. The HCl solution was standardized using NaHCO₃ as a primary standard.

A stock solution of TMA (1300 μ g/mL) was prepared from the 3.5M TMA solution and diluted quantitatively to make a series of standards (650, 325, 162.5, 81.3, 40.6, 20.3, 10.2, 5.1, 2.5, and 1.3 μ g/mL). Aliquots of each standard (200 μ L) were transferred to Eppendorf tubes, and the intensity of the m/z 60 ion generated from the headspace of each sample was monitored by an SIR experiment conducted under HePI-MS conditions. A standard curve was generated by plotting the intensity of

Table 1 TMA amounts found in some sushi and squid samples

Sample	Average amount (µg/g)
Squid sample (0.8 g)	17.2
A piece of salmon recovered from a sushi sample (0.8 g)	~ 0.24 ^a
A piece of shrimp recovered from a sushi sample (0.8 g)	~ 0.16 ^a

^aAmounts estimated by extrapolating the calibration curve

the m/z 60 ion peak in six replicates, against the amount of TMA present in each sample.

Estimating the freshness of seafood samples by the amount of TMA detected

Samples from shrimp- and salmon-sushi, as well as squid were analyzed in duplicate, immediately after they were brought to the laboratory, by an SIR experiment. The peak intensity at m/z 60 was monitored for 0.5 min for each sample (dwell time 0.1 s). Then, the average m/z 60 peak intensity was calculated for the total acquisition period of 0.5 min.

Effect of lime juice on TMA levels

After recoding positive-ion spectra (m/z 25–200) for 0.7 min, a 0.8-g sample of shrimp which had been kept at room temperature for 48 h was placed in the ion source. The volatiles emanating from the sample were monitored by recording a chronogram. After recording spectra for 1.2 min, a 500- μ L aliquot of lime juice was added,

using a pipette, to the shrimp sample, and spectra were acquired for another 1 min.

Results and discussion

Initial studies conducted with seafood samples showed that a peak at m/z 60 for protonated TMA can be observed under HePI-MS conditions. The intensity of the peak was insignificant in the spectra recorded from the headspace volatiles of fresh shrimp samples (Fig. 2a). However, a dramatic increase of its intensity (and that of the peak at m/z 61 for the protonated $^{13}C_1$ -isotopologue of TMA) was noted in the spectra recorded from samples kept at room temperature for 3 days (Fig. 2c). Throughout the experiments, the relative intensity ratios of the m/z 60 and 61 peaks were monitored to ascertain the integrity of the recorded signal as being due to TMA.

In a study that monitored the TMA levels over a period of 4 days, we noted that the released amount of TMA increased gradually as the samples aged (Fig. 3). The increase of the m/z 60 levels could be attributed to

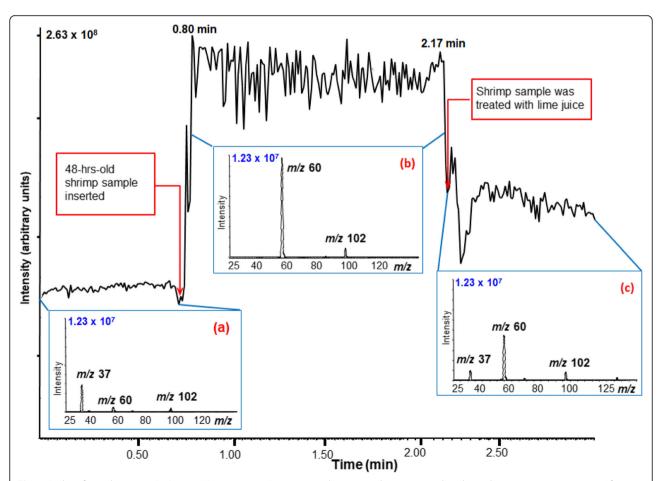


Fig. 5 A plot of signal intensity (m/z 25–200) versus time. At 0.7 min, a shrimp sample was inserted to the ambient-pressure ion source. After recoding spectra for 1.2 min, the sample was treated with lime juice. Panels **a, b,** and **c** show average mass spectra recorded before adding the sample, after adding the sample, and after adding lime juice, respectively

an increase of TMA because the m/z 61 peak showed a proportionate increase in intensity. Subsequently, in a comparative study, the average intensities of the m/z 60 peak were calculated for each seafood sample kept at room temperature for different periods of time and plotted against the age of the sample. The results showed that the rate of increase of TMA levels for squid samples kept at room temperature was higher than that for shrimp samples. For squid, the TMA amounts in headspace volatiles reached levels beyond the dynamic range of the method within 48 h at room temperature (Fig. 3b). In contrast, the TMA levels in the shrimp samples showed a gradual increase over a period of 96 h. Interestingly, the increase of TMA levels over time was relatively slower for fresh salmon and char samples (Fig. 3c, d). On the other hand, TMA emanating from cod reached a plateau level in 24 h (Fig. 3e).

To evaluate the linear dynamic range of quantification of the described method, a calibration plot was constructed by placing different amounts of TMA in the ion source. Figure 4 shows that intensity values increase linearly at least up to 16 μg of TMA. Using this calibration curve, we estimated the amount of TMA released from a sample of squid or samples of salmon and shrimp removed from sushi rolls (Table 1). Evidently, this is a semi-quantitative method at best because the absolute amount of TMA present in unit amounts of seafood or fish samples was not determined in the current study.

A major advantage of ambient-ionization mass spectrometric methods of analysis is specifically the fact that signals for analytes of interest can be elicited even from very complex samples without any measurable matrix interference—TMA is a gas. Once it emanates from a sample, it can be ionized and detected very efficiently because unlike in electrospray ionization, it does not have to be desorbed from a solution. In the samples used in this study, the matrix consists mostly of fat and protein. Unlike in electrospray or MALDI techniques, fats and proteins do not undergo ionization directly by HePI, and therefore do not interfere with TMA signals.

Masking of TMA odor with lime juice

In many parts of the world, it is customary for seafood to be served along with an acidic condiment such as lemon juice or lime juice (citric acid), vinegar (acetic acid), or tartar sauce (tartaric acid). To demonstrate the reduction of the TMA levels by the addition of lime juice, the source background signals were recorded for a period of 0.7 min. When a 48-h-old shrimp sample was introduced, a dramatic increase of the intensity of m/z 60 signal was noted (Fig. 5). The signal increase is caused by TMA accumulated in the headspace during the decomposition process. After the addition of lime juice, however, the signal abruptly dropped back to

background levels. This is due to the acid-base reaction that takes place between the basic TMA in seafood and lime juice or other acidic condiments (e.g., lemon or tartar juice), which convert TMA to its much less volatile respective salt (Fig. 5).

Conclusions

The described method can be used to rapidly screen the quality of seafood in a high-throughput manner, due to the simplified sample preparation procedure, which does not involve the solvent extraction of the analyte. The amount of TMA present can be determined semi-quantitatively. Herein, we have demonstrated that the reducing or completely eliminating the malodor associated with decaying seafood in seafood samples, by treating seafood with lime juice or vinegar, is due to reducing the amount of free TMA.

Abbreviations

HePI-MS: Helium-plasma ionization mass spectrometry; SIR: Selected-ion recording; TMA: Trimethylamine; TMAO: Trimethylamine *N*-oxide; UN: United Nations

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Authors' contributions

IH and TO carried out the experiments and collected, analyzed, and interpreted the experimental results. JP and ABA supervised the work and interpreted the experimental results. All authors contributed to the manuscript drafts, and read and approved the final manuscript.

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Availability of data and materials

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Competing interests

The authors declare that they have no competing interests.

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